



# THE MOMBASA POLYTECHNIC UNIVERSITY COLLEGE

(A Constituent College of JKUAT)

(A Centre of Excellence)

# Faculty of Engineering & Technology

DEPARTMENT OF BUILDING & CIVIL ENGINEERING

**CERTIFICATE IN BUILDING & CONSTRUCTION** 

EBC 1107: CHEMISTRY

END OF SEMESTER EXAMINATION SERIES: AUGUST 2012 TIME: 2 HOURS

### **Instructions to Candidates:**

You should have the following for this examination

- Answer Booklet

This paper consists of **FIVE** questions. Answer any **THREE** questions Maximum marks for each part of a question are as shown This paper consists of **THREE** printed pages

## **Question One (30 Marks)**

a)	Describe the structure of an atom.	(4 marks)
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- b) Define the following terms:
  - i) Passivity
  - ii) Polarization (4 marks)
- c) State **FIVE** control measures used to minimize galuamic corrosion. (5 marks)
- d) State any SIX constituents of a paint (3 marks)
- e) Explain how you can prevent failure of paint films. (4 marks)
- f) (i) Define the term polymers (2 marks) (ii) Using **TWO** examples in each case, state **TWO** types of polymers. (4 marks)
- g) State any **FOUR** types of chemical solutions. (4 marks)

## Question Two (15 marks)

Describe how the rate of corrosion in the plaint is affected by the following:

- i) Temperature
- ii) Water velocity
- iii) Oxygen
- iv) PH
- v) Condition and Composition of the metal surface

## **Question Three (15 marks)**

- a) State **TEN** characteristics of good paint (5 marks)
- b) Differentiate between thermoplastic and thermosetting plastics (5 marks)
- c) Explain the application of radiation in machine (5 marks)

#### **Question Four (15 marks)**

a) Describe the effect of passivity and polarization on the corrosion process.	(6 marks)
b) State eight characteristics of polymers	(4 marks)
c) State <b>THREE</b> types of radio actives	(3 marks)
d) Differentiate between weak acid and strong acid.	(2 marks)

## **Question Five (15 marks)**

- a) Describe the periodic trend of:
  - i) Electro negativity
  - ii) Ionization
  - iii) Meeting point

iv) Atomic radius (10 marks)

b) Down the group. (5 marks)

c) Across the period giving reasons for your answer. (5 marks)