

DEPARTMENT OF **PURE AND APPLIED SCIENCES** DIPLOMA IN SCIENCE LABORATORY TECHNOLOGY (DSLT 12S)

ACH 2320 : CHEMICAL ANALYTICAL TECHNIQUES III

SUPPLEMENTARY/SPECIAL : EXAMINATIONS

SERIES: MARCH 2014 TIME: 2 HOURS

INSTRUCTIONS:

You should have the following for this paper - Answer booklet This paper consists of *FIVE* questions. Answer Question **ONE** (compulsory) and any other **TWO** questions *This paper consists of 3 PRINTED pages*

QUESTION ONE

a) Diff	a) Differentiate co-precipitation and surface adsorption		
b) Exp	lain how degree of sugar satisfaction affect precipitation rate	(4 marks)	
c) Diff	erentiate between power compensated use and Heat flux use	(4 marks)	
d) Stat	e four properties of an ideal precipitations agent	(4 marks)	
e) Out	(4 marks)		
f) Des	cribe decantation procedure as a method of separation	(4 marks)	
g) Define the following terms:			
(i)	Glass transition temperature	(2 marks)	
(ii)	Thermal transition	(2 marks)	
(iii)	First order transition	(2 marks)	

QUESTION TWO

a)	(i)	State two types of ion-exchange membrane	(2 marks)
	(ii)	State four properties of ion-exchange membrane	(4 marks)
b)	Expla	in the procedure for soxlet extraction	(5 marks)

QUESTION THREE

Explain any five factors affecting formation of precipitate	(15 marks)
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QUESTION FOUR

a) Explain the gravimetric analysis procedures	(7 marks)
b) An iron ore was analyzed by dissolving 1.1325 sample in concentrated	l HCl, the resulting
solution was diluted with water and ion III was precipitated as hydrous oxide	e Fe ₂ O ₃ .xH ₂ O, which
was dehydrated to give 0.539g of pure Fe ₂ O ₃ (159.69g/mol)	
Calculate	

(i)The percentage of iron(4 marks)(ii)Percentage of Fe_2O_3 in the sample(4 marks)

QUESTION FIVE

a)	Describe the principle and procedure of affinity chromatography.	(10 marks)
b)	State five properties of ulcer membrane for reverse osmosis application	(5 marks)