

Faculty of Applied and Health Sciences

DEPARTMENT OF PURE AND APPLIED SCIENCES

DIPLOMA IN ANALYTICAL CHEMISTRY (DAC 11M)

ACH 2307: ORGANIC CHEMISTRY III

SEMESTER: EXAMINATIONS

SERIES: DECEMBER 2013

TIME: 2 HOURS

INSTRUCTIONS:

You should have the following for this paper - Answer booklet

This paper consists of *FIVE* questions.

Answer Question ONE (compulsory) and any other TWO questions

This paper consists of 3 PRINTED pages

Question ONE

a)	Giving	Giving an example, explain the meaning of the following terms used in polymer industries.		
	(i)	Homopolymer		
	(ii)	Copolymer		
	(iii)	Addition polymer		
	(iv)	Condensation polymer		
	(v)	Monomer		
			(10marks)	
b)	State	FIVE advantages of emulsion polymerization	(10marks)	
c)	With a	in example in each case explain the formation of the following isomers.		
	(i)	Optical		
	(ii)	Geometrical		
	(iii)	Functional group		
			(10marks)	
Question TWO				
a)	Using	Using an example explain the meaning of the following terms		
	(i)	Tautomerism	(4marks)	
	(ii)	Chiral carbon	(2marks)	
b)) Sec butyl chloride has two stereoisomeric forms. Draw their mirror images (optical ison			
and show how		now how they rotate plane polarized light. Also indicate which one is	dextrorotary and	
	levoro	tatory.	(6marks)	
c) Give an example of a mo		n example of a monomer with		
	(i)	Functionality of one		
	(ii)	Functionality of two		
	(iii)	Functionality of three		
			(6marks)	
d)	List th	ne main reactants that are used in solution polymerization	(2marks)	

Question THREE

a) State FIVE difference between thermoplastics and thermosetting plastics. (10marks)
b) List seven draw backs of rubber. (7marks)
c) State TWO disadvantage of solution polymerization process (3marks)

Question FOUR

a) On a polymer state verses temperature graph indicate different polymer states Tg and Tm points.

(5marks)

b) List and discuss any five factors that affects Tg. Value of a polymer (15marks)

Question FIVE

- a) (i) Define a condensation polymer. (2marks)
 - (ii) Naming the monomers, write condensation polymerization reaction for formation of nylon 66. (6marks)
- b) State SIX advantages of valcanised rubber. (12marks)