



Technical University of Mombasa

Faculty of Applied and Health Sciences

DEPARTMENT OF PURE AND APPLIED SCIENCES

DIPLOMA IN ANALYTICAL CHEMISTRY
(DAC 12S)

ACH 2207: CHEMISTRY OF AROMATIC COMPOUNDS

SEMESTER: EXAMINATIONS

SERIES: DECEMBER 2013

TIME: 2 HOURS

INSTRUCTIONS:

You should have the following for this paper

- *Answer booklet*

This paper consists of **FIVE** questions.

Answer Question **ONE (compulsory)** and any other **TWO** questions

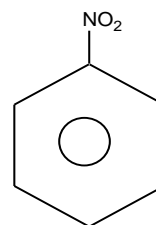
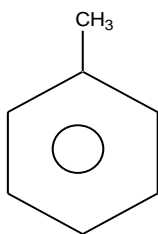
This paper consists of 5 PRINTED pages

Question ONE

- a) Give the IUPAC names of the following
- b) Explain the meaning of the following terms giving examples in each case.
- (i) Activating group (2marks)
 - (ii) Deactivating group (2marks)
 - (iii) Ortho – directing (2marks)
- c) Outline any THREE characteristic features of Aromatic compounds (6marks)
- d) I. Draw the structures of the following compounds
- (i) Naphalene
 - (ii) Anthracene
 - (iii) Phenanthrene
 - (iv) 3-Nitronaphthalene
 - (v) 2-Naphthol
 - (vi) 1,4 – Naphthoquinone
 - (vii) Quinone
- (7marks)
- II. Distinguish between the following
- (i) Electrophite and Nucleophile (2marks)
 - (ii) Homolytic and Heterolytic fission (2marks)
 - (iii) Aromatic and Anti-aromatic compound (2marks)

Question TWO

- a) Outline any FIVE properties of benzene (5marks)
- b) (i) State any TWO reagents needed for nitration of benzene (1mark)
- (ii) Outline the mechanism for the nitration of benzene. (5marks)
- (iii) State the role of Conc. H_2SO_4 in the reaction (1mark)
- c) Which of the two compounds below is more reactive. Give reason(s) for your answer.



Question THREE

a) Define the following

(i) Fiedel-crafts Alkylation

(1 ½ marks)

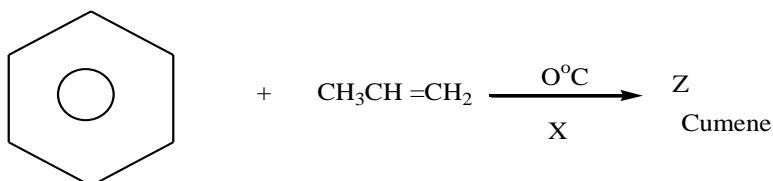
(ii) Lewis acid

(1mark)

(iii) Acylation

(½ mark)

b) Complete the following equation by naming X and drawing the structure of Z



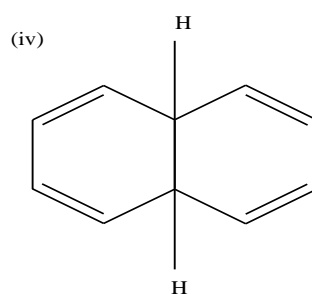
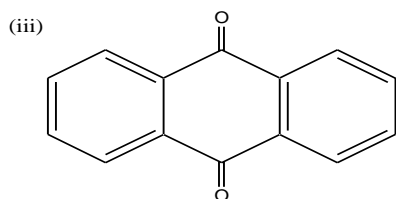
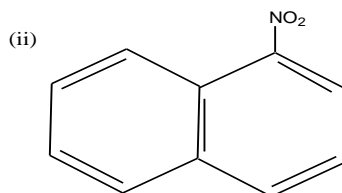
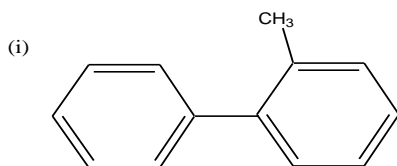
(2marks)

Question FOUR

a) Outline TWO limitations of Friedel crafts alkylation.

(2marks)

b) Name the following compounds

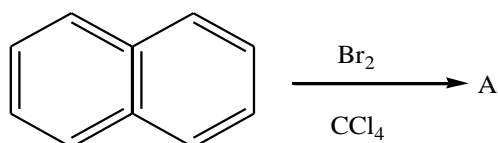


(4marks)

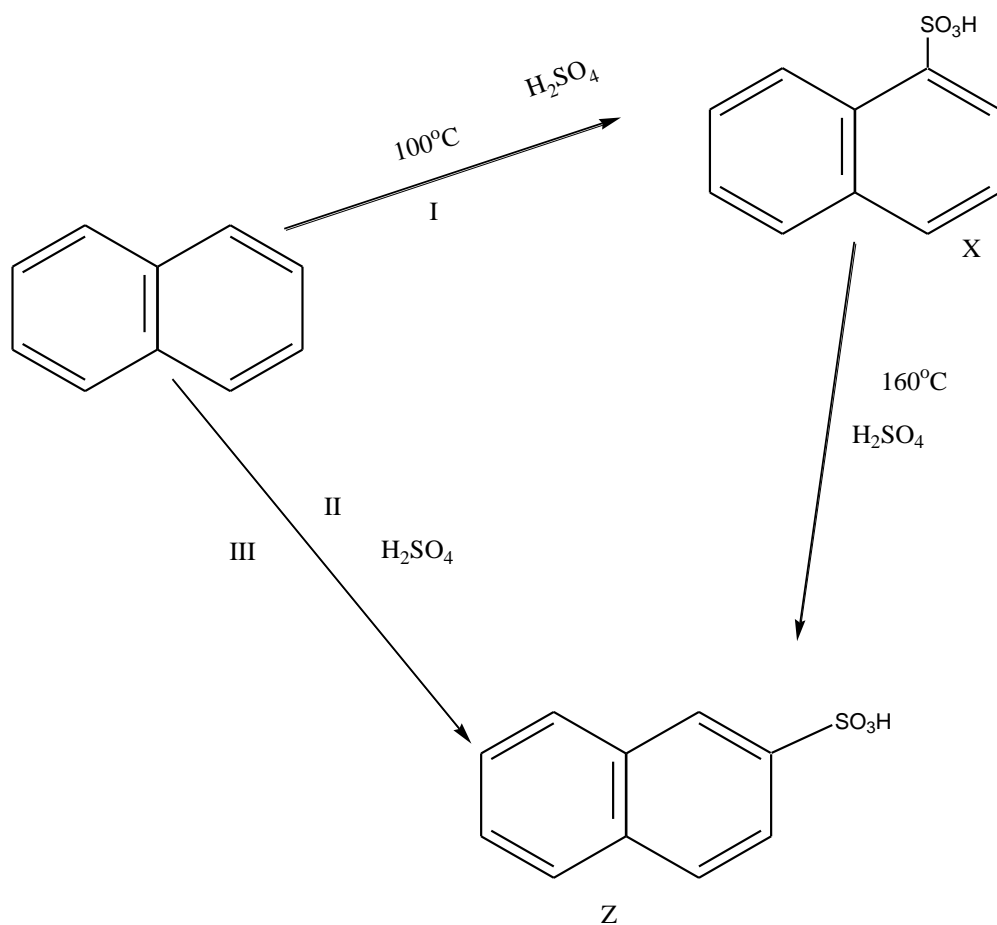
c) Give reason(s) why biphenyl are move reactive than benzene.

(2marks)

d) Complete the following equations



e) Study the reaction



- (i) Name the conditions I, II and III (3marks)
- (ii) Name the resulting compounds X and Z (3marks)

Question FIVE

a) What product is formed when benzene is treated with each organic halide in the presence of AlCl_3

(i) $(\text{CH}_3)_2\text{CHCl}$ (1mark)

(ii)  (1mark)

(iii) $\text{CH}_3\text{CH}_2\text{C}(=\text{O})\text{CH}_3$ (1mark)

b) Write a mechanism for the product formed in

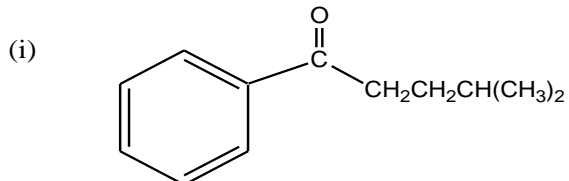
(i) a(i)

(3marks)

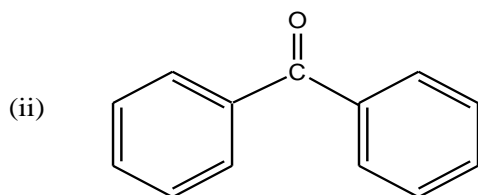
(ii) a(ii)

(3marks)

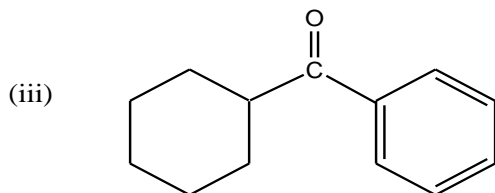
c) What acid chloride would be needed to prepare each of the following ketones from benzene using friedel –crafts acylation



(2marks)



(2marks)



(2marks)