



Technical University of Mombasa

Faculty of Applied and Health Sciences

DEPARTMENT OF PURE AND APPLIED SCIENCES

DIPLOMA IN ANALYTICAL CHEMISTRY

(DAC 11M)

ABT 2308 : BIOCHEMICAL TECHNIQUE OF ANALYSIS

SEMESTER: EXAMINATIONS

SERIES: DECEMBER 2013

TIME: 2 HOURS

INSTRUCTIONS:

You should have the following for this paper

- *Answer booklet*

This paper consists of **FIVE** questions.

Answer Question **ONE (compulsory)** and any other **TWO** questions

This paper consists of 2 PRINTED pages

Question ONE

- (a) Distinguish between lyases and ligase (4marks)
- (b) Define the following term
- (i) Buffer solution (2marks)
 - (ii) Enzyme (2marks)
- (c) Calculate the pH of a solution that has $[H^+]$ of $1 \times 10^{-5} \text{ mol/l}$ (4marks)
- (d) State FOUR disadvantage of protein concentration by precipitation technique (4marks)
- (e) State any FOUR ligands, used in affinity chromatomply (4marks)
- (f) State any SIX limitation of Beer-Lambert laws (6marks)
- (g) What are the symptoms of aged measuring electrode (4marks)

Question TWO

- (a) State the three major buffer found in human body (3marks)
- (b) Briefly explain the role of buffer mentioned above (12marks)

Question THREE

- (a) Describe the principle and procedure of Affinity chromatography (12marks)
- (b) What are suicidal inhibitor? Give example (3marks)

Question FOUR

- (a) Buffer solution is made by adding 328g CH_3COONa salt to 1 dm^3 of 0.01m CH_3COOH what is the pH of the buffer when K_a of CH_3COOH is $1.7 \times 10^{-5} \text{ mol}^{-1} \text{ dm}^{-3}$ (9marks)
- (b) If 25ml of 0.1mHCl solution is diluted to 100ml with distilled water calculate pH of resulting solution (6marks)

Question FIVE

- (a) Explain how the following agent induces protein precipitation
- (i) Neutral salt (4marks)
 - (ii) Organic solvent (4marks)
 - (iii) High molecular weight polymer (4marks)
- (b) Explain why biospecific Affinity chromatography is not commonly used in industries (3marks)