



Technical University of Mombasa

Faculty of Applied and Health Sciences

DEPARTMENT OF **PURE AND APPLIED SCIENCES**

DIPLOMA IN ANALYTICAL CHEMISTRY

(DAC 10J)

ACH 2318 : INSTRUMENTAL METHODS OF ANALYSIS

SPECIAL/SUPPLEMENTARY: EXAMINATIONS

SERIES: FEBRUARY 2013

TIME: 2 HOURS

INSTRUCTIONS:

You should have the following for this paper

- *Answer booklet*

This paper consists of **FIVE** questions.

Answer Question **ONE (compulsory)** and any other **TWO** questions

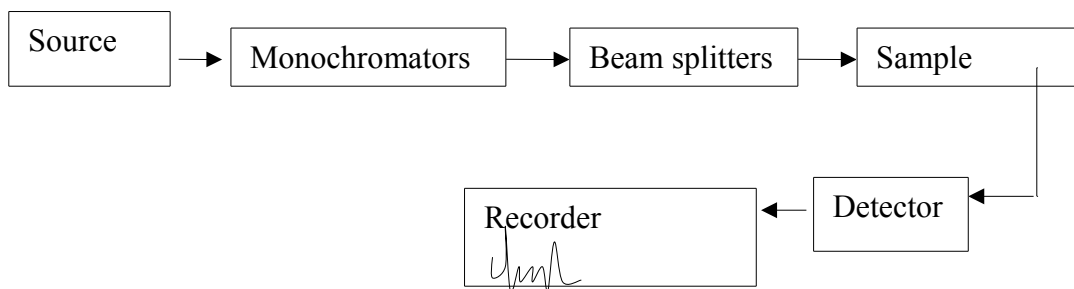
This paper consists of 4 PRINTED pages

Question ONE

- a) (i) State and explain THREE factors which affect the efficiency of a chromatographic column **(3marks)**
- (ii) Give FOUR factors influencing vibrational frequencies in IR spectroscopy. **(2marks)**
- (iii) Explain briefly the nature of TWO different types of flames used in AAS. **(3marks)**
- b) (i) Give THREE advantages why using narrower a wavelength band is preferred in spectroscopic analysis. **(3marks)**
- (ii) Define the following terms as used in UV – visible spectrophotometry. **(4marks)**
- I. Auxochrome
 - II. Chromophore
 - III. Hypochromic effect
 - IV. Bathochromic effect
- (iii) How many fundamental vibrational frequencies would you expect to observe in the infrared absorption spectrum of CO₂. Justify your answer. **(5marks)**
- c) (i) Give the difference between normal phase and reversed phase chromatographic techniques. **(4marks)**
- (ii) Explain THREE reasons why temperature control is important in HPLC and LC analysis. **(3marks)**
- (iii) State THREE functions performed by power supply system in UV spectrophotometers. **(3marks)**

Question TWO

- a) (i) Describe briefly the working principles of a double beam spectrophotometer (Use a block diagram). **(5marks)**



- (ii) State THREE advantages of a double beam spectrophotometer over a single beam spectrophotometer. **(3marks)**
- (iii) Define the following terms **(4marks)**
- (I) Releasing agent
 - (II) Protecting agent

- (III) Dead time
 (IV) Attenuation
- b) (i) Calculate the approximate frequency of the C-H stretching vibration from the following data.
- K = $500\text{Nm}^{-1} = 5.0 \times 10^5\text{gS}^{-1}$ (since $1\text{N} = 10^3\text{gm}^{-1}$)
 Mc = Mass of carbon atom = $2.0 \times 10^{-24}\text{g}$
 MH = Mass of Hydrogen atom = $1.6 \times 10^{-24}\text{g}$
 C = $3.0 \times 10^8\text{ms}^{-1}$

(3marks)

Question THREE

- a) (i) Define :
- I. Monochromator
 - II. Noise
- (ii) State FOUR methods used to prepare solid samples for IR analysis
- b) (i) State THREE causes of deviation from Beer-Lambert Law
- (ii) Identify THREE types of chemical reactions occurring in the flame in FES
- (iii) List FOUR types of Nebulization
- c) A stock solution was made by dissolving 0.900g of Potassium permanganate (KMnO_4) in water and diluting to 250cm^3 . Calculate the concentration of the solution in ppm with respect to manganese (RFM of $\text{KMnO}_4 = 158.036$)

(2marks)

(2marks)

(3marks)

(3marks)

(2marks)

(3marks)

Question FOUR

- a) (i) Describe the working principles of a phototube detector
- (ii) Name TWO types of filters and TWO types of monochromators.
- (iii) Give THREE requirements of a radiation sources.
- b) Describe determination of the formula of complexes by Job's method

(5marks)

(2marks)

(3marks)

(5marks)

Question FIVE

- a) (i) A solution containing TWO substances P and Q has an absorbance in 1cm cuvette of 0.36 at 350nm and 0.225 at 400nm. The molar absorption coefficients of P and Q at this wavelength are given below

(5marks)

Compound	Wavelength(μl)	
	At 350nm	At 400nm
P	1500	3000
Q	7000	6500

Calculate the concentrations of P and Q

- (ii) Identify all symbols in the expression (2marks)

$$\mu = \frac{M_1 M_2}{M_1 + M_2}$$

- (iii) Frequency of vibration is given by the equation

$$v = \frac{1}{2\pi c} \sqrt{\frac{K}{\mu}}$$

Identify the symbols v, c and k and state one physical significance of f . (2marks)

- (b) (i) A column with height 100cm gave the following experimental results.

Trial(no.)	Flow rate cms^{-1}	tRSec	W(sec)
1	1.0	300	80
2	2.5	100	24
3	5.0	50	10

Calculate the number of theoretical plates N and the Height equivalent of a theoretical plate (HETP) (3marks)

- (ii) State the physical significance of number theoretical plates (N) and the height equivalent of a theoretical plate H. in HPLC.

(2marks)

- (iii) Give One application of IR in the industry (1mark)