

Technical University of Mombasa

Faculty of Applied and Health Sciences

DEPARTMENT OF PURE AND APPLIED SCIENCES

DIPLOMA IN ANALYTICAL CHEMISTRY (DAC 11M)

ACH 2207: CHEMISTRY OF AROMATIC COMPOUNDS

SPECIAL/SUPPLEMENTARY: EXAMINATIONS

SERIES: February 2013

TIME: 2 HOURS

INSTRUCTIONS:

You should have the following for this paper

- Answer booklet

This paper consists of *FIVE* questions.

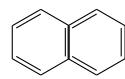
Answer Question ONE (compulsory) and any other TWO questions

This paper consists of 4 PRINTED pages

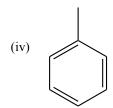
Question ONE

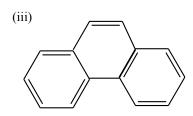
a) Give IUPAC names for the following

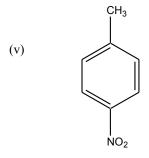




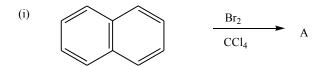


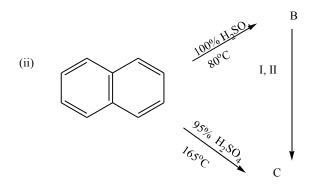






b) Complete the following equations





- (iv) The conditions I and II (2marks)
- c) Explain the meaning of the following
 - (i) Electrophile (1mark)
 - (ii) Nucleophile (1mark)
 - (iii) Biphenyls (1 ½ marks)
 - (iv) Friedel crafts Acylation (2marks)
- d) (i) Outline the mechanism for (i) Friedelcrafts alylation of benzene using ethanoyl chloride and AlCl₃ at 80°C (6marks)
 - (ii) Friedel crafts alkylation of Benzene with 2-chloropropane and AlCl₃ as a halogen career.

(6marks)

e) State any TWO applications of benzene derivatives

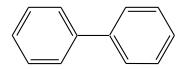
(1mark)

Question TWO

- a) Distinguish between activating and deactivating groups (4marks)
- b) Name FOUR activation of groups and arrange then in order of their activating power and explain this order. (4marks)
- c) Give reason(s) why benzene is said to be stable. (2marks)
- d) Outline TWO physical properties of benzene (1mark)

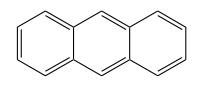
Ouestion THREE

a) Name the following compounds



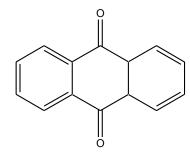
В

C



D

Е



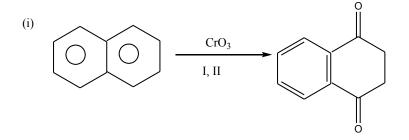
(5marks)

b) Explain the following

- I. A is more reactive to electrophilic aromatic substitution than benzene. (2marks)
- II. Write equations for the following
 - (i) Bronminatin of naphthalene in presence of CCl₄

(2marks)

- (ii) State the other condition required for this reaction
- (1mark)
- III. (i) Complete the following equations giving the appropriate names



$$\begin{array}{c} & \\ \hline \\ & \\ V_2O_5 \end{array} \qquad B$$

D
$$H_2$$
, Pt H_2 /Ni H_2 , Pd H_2 , Pd Pressure, heat

(2marks)

Question FOUR

a) Outline the limitations of Friedel crafts alkylation and acylation (10marks)

b) Explain the synthetic applications Friedel-Crafts alkylation (5marks)

Question FIVE

- a) Draw the structures of the following organic compounds
 - (i) Thiophene
 - (ii) Pyridine
 - (iii) Pyrole
 - (iv) Furan
 - (v) Tetrahydro furan

(5marks)

- b) Write two equations to show the effect of substituent groups in the benzene ring (4marks)
- c) (i) State TWO Lewis acids commonly used in friedel –crafts alkylation (1mark)
 - (ii) Describe the general mechanism for electrophilic Aromatic substitution (5marks)