

### TECHNICAL UNIVERSITY OF MOMBASA

Faculty of Engineering and Technology
Department of Mechanical & Automotive Engineering
UNIVERSITY EXAMINATION FOR:
Diploma in Mechanical Engineering
APS 2150: Physical Science for Engineers
END OF SEMESTER EXAMINATION
SERIES: AUGUST 2019
TIME: 2 HOURS
DATE:

# **Instruction to Candidates:**

You should have the following for this examination

Answer booklet

This paper consists of **FIVE** questions. Attempt any THREE questions.

Maximum marks for each part of a question are as shown.

Do not write on the question paper.

#### **Question ONE**

a) State any SIX similar properties of sound and light waves.	(6 marks)
b) Using sketches, explain how the eye can see objects.	(6 marks)
c) Explain any FOUR laws of optics and state an application for each law.	(8 marks)

# **Question TWO**

a)	) Explai	n the T	HREE mod	des of h	neat	transfer	giving a	n example	for eac	h. <b>(</b> 6	marks)
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- b) Using a change of state graph, show the change of state of water from -10 °C to 300 °C showing all the important temperature points, states and latent heats. (10 marks)
- c) State any FOUR heat transfer laws giving an application for each. (4 marks)

### **Question THREE**

a) Explain the term <i>Thermodynamics</i> and explain its first law.	(4 marks)
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- b) State any SIX properties of an ideal gas or fluid. (6 marks)
- c) State any FIVE thermodynamic fluids used in engineering applications. (5 marks)
- d) State any SIX applications of thermodynamics in engineering applications. (6 marks)

# **Question FOUR**

- a) Explain any FIVE organic compounds stating an application for each. (10 marks)
- b) Explain the term "Thermo-chemistry" and define the principle of electrolysis stating two applications. (10 marks)

# **Question FIVE**

- a) Explain how the periodic table of elements is arranged in terms of reactivity, physical and chemical properties. (10 marks)
- b) Explain any THREE types of bonding of elements.

(6 marks)

c) Define the following terms giving an example for each;

(4 marks)

- i. Mixture
- ii. Element
- iii. Compound
- iv. Molecule