

TECHNICAL UNIVERSITY OF MOMBASA

FACULTY OF APPLIED AND HEALTH SCIENCES DEPARTMENT OF MEDICAL SCIENCES UNIVERSITY EXAMINATION FOR:

DIPLOMA IN PHARMACEUTICAL TECHNOLOGY

APM 2316 : ORGANIC & INORGANIC PHARMACEUTICAL CHEMISTRY
END OF SEMESTER EXAMINATION

SERIES:AUGUST2019

TIME:2 HOURS

DATE: Pick Date Aug 2019

Instructions to Candidates

You should have the following for this examination

-Answer Booklet, examination pass and student ID

This paper consists of **THREE**Section(s). Attempt All questions in section A and B and any two questions in section C.

Circle the correct answer in section A.

SECTION A (40 MARKS)

Answer all the questions

- 1. Dietary supplementation of calcium becomes necessary during the following EXCEPT
 - A. Pregnancy
 - B. Lactation
 - C. Period of rapid growth
 - D. Acromegaly
- 2. Ascorbic acid is normally administered with haematinics in order to:
 - A. Convert Fe²⁺ to Fe³⁺
 - B. Convert Fe³⁺ to Fe²⁺
 - C. Reduce pH of intestinal fluid
 - D. None of the above
- 3. Choose the incorrect statement in regard to non-systemic antacids
 - A. Their antacids effect is localized to the gastro-intestinal tract
 - B. The compounds used are generally soluble in the GIT fluid and plasma
 - C. Examples of cations used include aluminium magnesium and calcium bismuth
 - D. Anions used include carbonates, phosphates hydroxides or silicates
- 4. Which of the trace element is involved as a coenzyme in protein synthesis and is a central feature of Vit .B?
 - A. Molybdenum
 - B. Magnesium
 - C. Zinc
 - D. Cobalt
- 5. Fe³⁺(ferric) employed in medicinal products is frequently encountered in the form of one of its soluble complex below EXCEPT
 - A. Ferric pyrophosphate
 - B. Ferric ammonium tartarate
 - C. Ferric ammonium citrate
 - D. Ferric iodide

- 6. The following are uses of talc EXCEPT
 - A. Used in dusting powders as protective's
 - B. Used in medicated dust in topical products
 - C. Used as filter aid
 - D. Used as a suspending agent
- 7. Choose the incorrect statement in regard to calcium gluconate solution:
 - A. It is less irritating compared to calcium chloride
 - B. It is extensively used as a calcium supplement
 - C. It can be administered either via intramuscular or intravenous route
 - D. It is stabilized by protein lecithin as an injection
- 8. The following statements are FALSE in respect to calcium ion. Choose the correct one:
 - A. About 99% of body calcium is normally found in the liver
 - B. Diffusible calcium in the body is plasma bound
 - C. Phosphate ion is normally an ion that interferes with calcium in the body inversely
 - D. Proper calcium absorption requires adequate levels of Vit. D.
- 9. Which of the following anions is not formulated for systemic use?
 - A. Acetate
 - B. Bicarbonate
 - C. Carbonate
 - D. Citrate
- 10. Which of the following statements about $[Fe(H_2O)_6]^{3+}$ is correct
 - A. Readily undergo hydrolysis
 - B. Is readily absorbed in the duodenum
 - C. Readily undergoes oxidation in acidic media
 - D. A and B are correct
- 11. Peroxides of Mg may be obtained from
 - A. Precipitation of sea water
 - B. Treatment of Mg(OH)₂ with hydrogen peroxide
 - C. Heating MgO at high temperature
 - D. Dehydrating $Mg(OH)_2$

- 12. Which of the following is NOT used as internal protective?
 - A. Aluminum hydroxide
 - B. Medicinal Zinc peroxide
 - C. Bismuth subnitrate
 - D. Magnesium trisilicate
- 13. In REDOX reaction, the oxidizing agent
 - a) Is oxidized
 - b) Loses electrons
 - c) Is reduced
 - d) Increased in oxidation state
- 14 Which of the following statement is correct regarding the mode of action of metformin?
 - a) It stimulates insulin release from the beta cells
 - b) It increases glucose utilization in the muscles
 - c) It enhances gluconeogenesis
 - d) It promotes secretion of insulin from beta cells

Figure 1 below represents a class of drugs used in management of hyperglycaemia. Use it to answer the following two questions

$$R \xrightarrow{\bigcup_{i=1}^{N} \prod_{j=1}^{N} \prod_{i=1}^{N} \prod_{j=1}^{N} \prod_{j=1}^{N}$$

Figure 1

- 15 Which of the statements below is correct concerning the SAR of the above class of compounds?
 - a) Introduction of aromatic groups at R1 does not affect activity
 - b) A methyl substituent para to the sulfonyl aromatic ring gives potent compounds
 - c) A bulky group at R position abolishes activity
 - d) Bulky group at position R1 is not necessary for activity
- 16 Select the correct statement regarding the above class of compounds:
 - a) Their main metabolic pathway is aromatic ring hydroxylation
 - b) They act by enhancing insulin secretion
 - c) They are known as meglitinides

d) Enhance the insulin utilization at the receptors

17 Which of the following statements is correct regarding angiotensin converting enzyme inhibitors with a carboxylate group in the structure?

- a) They are associated with skin rash and taste disturbances
- b) They have good renal clearance properties
- c) They have high lipophilicity
- d) They are ultra-short acting

18 Select the correct alternative about angiotensin receptor blockers;

- a) They block the angiotensin I receptors
- b) They are formulated as acidic salts
- c) They are formulated as basic salts
- d) They are formulated as ester pro-drugs

19 Select the correct statement regarding the structure of cardiac glycosides:

- a) Common sugars linked to the nucleus include L-glucose and L-digitoxose
- b) They consist of a 5-membered lactone ring as part of the structure
- c) The aglycone portion carries the sugar and the steroid nucleus
- d) The non-sugar portion is known as the gennin

20 Which of the following CVS agents is classified as a phenyalkylamine derivative?

- a) Verapamil
- b) Diltiazem
- c) Disopyramide
- d) Minoxidil
- 21 Which of the following statement is true concerning the structure-activity relationship of the angiotensin converting enzymes inhibitors whose structure is shown in figure 2?

- a) They should have an aromatic ring for activity
- b) Esterification decreases lipid solubility

- c) Presence of a carboxylic group increases side effects
- d) Those with a sulfhydryl group have a short duration of action

22 Which of the following statement is true regarding spironolactone?

- a) It is an agonist of aldosterone
- b) Its use can result in severe hyperkalemia
- c) Has a very high diuretic effect
- d) It blocks sodium channels in the distal convoluted tubule

23 Meclizine, an antihistamine has been derived from:

- a) Alkylamines
- b) Propylamines
- c) Ethylene diamines
- d) Ethanolamines

24 Which of the following statement is incorrect about heparin?

- a) It is a mucopolysaccharide
- b) It is a inhibits conversion of prothrombin to thrombin
- c) It consists of sulphated glucosamine and glucuronic acid units
- d) It has good oral bioavailability

25 Which of the following alternatives is correct regarding warfarin?

- a) It is a calcium sequestering agent
- b) It is a coumarin derivative
- c) It is suitable for emergency haemorrhage management
- d) It is chemically classified as an indandione

26 The main application of haemostatic agents is:

- a) To arrest bleeding
- b) Antiplatelet effect
- c) Fibrinolysis
- d) Blood storage

27 Short acting insulin is also known as:

- a) Soluble insulin
- b) Ultralente insulin
- c) Insulin lispro

- d) Insulin aspartic
- 28. Which of the following statements is true about precipitated chalk?
 - a) it is more reactive than the prepared chalk
 - b) it is the form of chalk used for preparation of chalk mixtures
 - c) its antacid capacity is much less than that of aluminium hydroxide
 - d) it is less reactive than the prepared chalk
- 29. Glass materials are generally prepared by fusing pure silica with?
 - a) sodium bicarbonate
 - b) calcium carbonate
 - c) sodium hydroxide
 - d) potassium chloride
- 30. The following statements regarding Bismuth compounds as gastric antacids are true except;
 - a) small amount of dissolved bismuthyl cations also exerts mild antiseptic action
 - b) the compounds owe their antacid action mostly due to the cations
 - c) the compounds have no constipatory effects
 - d) none of the above
- 31. Citric acid 1.0 % is normally added to magnesium hydroxide prepared by hydration for Use in preparation of gastric antacid. The purpose of this acid is to?
 - a) improve the attractiveness of the rather chalky appearance of the mixture
 - b) improve the effectiveness of the antacid
 - c) enhance the stability of the product during storage
 - d) choices B and D
- 32. Which of the following statements is not true of simethicone?
 - a) It is a polymeric dimethylsiloxane
 - b) It can be used an antiflatulant
 - c) It exhibits jellylike to solid like properties
 - d) It chemically consists of both carbon and silicon
- 33. Sodium chloride can be used in the following situations EXCEPT?
 - A. As an enema
 - B. As a plasma expander
 - C. As a urinary alkalinizer

- D. In treatment of exposed tissues
- 34. Which of the following statements is true about Fluoridation?
 - A. Involves formation of fluoride salts
 - B. Is a major cause of dental carries
 - C. Cannot be done domestically
 - D. Sodium fluorosilicate (Na2siF6) is the compound of choice for large scale fluoridation of drinking water
- 35. Ringer's lactate is contra-indicated in the following conditions except?
 - A. Lactic acidosis
 - B. Congestive heart failure
 - C. Mild metabolic acidosis
 - D. Renal impairment
- 36. Dextrose is used in replacement of water loss in the following conditions except?
 - A. Fevers
 - B. Diarrhea
 - C. Diabetes Melitus
 - D. Malaria
- 37. The following are TRUE on calcium requirements except?
 - A. Normal requirements range from 0.2- 1.0 g / day
 - B. Dietary requirement is lower in old age due to reduced metabolism
 - C. Dietary requirement is higher during pregnancy and lactation
 - D. The daily requirement is higher in childhood
- 38. Which of the following is false regarding Fluoride?
 - A. Ion is toxic
 - B. Is found in bones and teeth
 - C. Is given as NaF
 - D. Optimum level in drinking water is 2ppm
- 39. The following statements on iodine are true except?
 - A. Radioiodine is used for the treatment of hyperthyroidism
 - B. Tincture of iodine doesn't contain alcohol
 - C. Iodine deficiency can lead to mental retardation

- D. Iodine deficiency can cause enlargement of the thyroid gland
- 40. Which of the following statements on iodine supplements is not true?
 - A. Lugol's solution contains both potassium iodide and iodine
 - B. Iodide is converted to iodine in GIT
 - C. Iodized salt is a source of iodine
 - D. Ammonium iodide can be used as a supplement

SECTION B (40 MARKS)

Answer ALL questions from this section

- 41. Name 4 agents used as vasodilators in the control of hypertension (4 marks)
- 42. Outline general uses of dimethysiloxanes

(4marks)

- 43. List Four (4) groups of drugs used to treat arrhythmias giving a suitable example in each group

 4 marks
- 44. State four SAR of the thiazide diuretics whose structure is as shown in Figure below

4 marks

- 45. (a) Name 4 agents used as vasodilators in the control of hypertension
- 2 marks

(b) List four non-sedating antihistamines

- 2 marks
- 46. Differentiate between proton pump inhibitors and histamine-2 receptor antagonists in terms of mode of action. **4 marks**
- 47. What advantages do proton pump inhibitors have over histamine-2 receptor antagonists in therapeutics?

 4 marks
- 48. Classify the anticoagulants agents and give suitable examples in each class 4 marks
- 49. Explain how stains on skin caused by silver compounds are removed. Use equation(s) where necessary. (4marks)
- 50. Define fluorosis and describe how it occurs

(4 marks)

SECTION C (40 MARKS)

Answer ANY TWO questions from this section

51. Below is a structure for morphine. Use it to answer questions that follow.

$$\begin{array}{c} \text{CH}_3 \\ \text{10} \\ \text{9} \\ \text{N}^{17} \\ \text{16} \\ \text{11} \\ \text{B}^{14} \\ \text{12} \\ \text{13} \\ \text{C} \\ \text{OH} \\ \end{array}$$

- a) Explain its physicochemical properties 4 marks
- (b) Explain the structural modification of morphine structure that results in morphinans.

Giving an example and use of morphinans

4 marks

(c) State eight SARs of morphine

8 marks

(d) Explain the structural modifications on the morphine structure below that would

Result in narcotic antagonists

4 marks

52. (a) Explain with suitable examples types of glass of interest in pharmacy (10 marks)

(b) Discuss germicides in terms of

(10 marks)

- i. Selection of a suitable germicide
- ii. Different classes of germicides and their advantages and disadvantages
- 53. Below is a structure of angiotensin converting enzymes inhibitors?

$$\begin{array}{c|c} & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & &$$

(a)

(i) Briefly discuss their mode of action

2 marks

- (ii) Discuss their structure activity relationship (SAR) 8 marks
- (b) The structure below belongs to a class of drugs used in the management of hypertension. Use the structure to respond to the questions that follow.

- (i) Give the general class and the specific chemical class to which these drugs belong **4 marks**
- (ii) Discuss the group SAR 6 marks