

TECHNICAL UNIVERSITY OF MOMBASA

FACULTY OF APPLIED AND HEALTH SCIENCES DEPARTMENT OF PURE & APPLIED SCIENCES

UNIVERSITY EXAMINATION FOR:

DIPLOMA IN ANALYTICAL CHEMISTRY

ACH2205: ORGANIC CHEMISTRY II

END OF SEMESTER EXAMINATION

SERIES: DECEMBER 2016

TIME: 2 HOURS

DATE: Pick Date Select Month Pick Year

Instructions to Candidates

You should have the following for this examination

-Answer Booklet, examination pass and student ID

This paper consists of FIVE questions. Attempt question ONE (Compulsory) and any other TWO questions.

Do not write on the question paper.

Question ONE

(a) (i) Draw and name structures of examples of 1°,2° and 3° amines that contain four Sp³ hybridized carbon atoms.

(7 marks)

(ii) Using the same criterion name and provide examples of $\mathbf{1}^{0}$, $\mathbf{2}^{0}$ and $\mathbf{3}^{0}$ alcohols.

(6marks)

(iii) How does the classification system differ between the two functional groups.

(4marks)

(b) Give the disadvantage of using CCl₂F₂ as an aerosol propellant.

(3marks)

(c) A grignard reagent CH₃CH₂MgCl react with an aldehyde CH₃CH₂CHO and ketone

CH₃COCH₃ to form different products. Write equations for formation of these products

(6marks)

(d) Explain why carboxylic acids have higher boiling points than corresponding alkanes of

comparable mass. (4marks)

and name them.

(a) With help of equations, explain how you would obtain CH ₃ CH ₂ COOH from an alkylhalide. (Hint:-it's a two steps reaction). (7marks) (b) Both aldehyde and ketone have polar carbonyl group in their molecules but aldehyde is readily oxidized while ketone is not. (i) Give a reason for this kind of behavior. (3marks) (ii) Name the product of this oxidation. (1marks) (c) Describe a simple chemical test of identifying aldehyde and ketone in the lab. (4marks) Question THREE Account for the following behaviors. (i) Acyl halide undergo nucleophilic substitution reaction more readily than alkyl halides. (7marks) (ii) Ethanamide (CH ₃ CONH ₂) is less nucleophilic than Ethylamine (CH ₃ CH ₂ NH ₂) (3marks) (iii) Carboxylic acids do not undergo electrophilic or nucleophilic substitution. (5marks)
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(a) State (i) FOUR physical properties of carboxylic acids. (10marks)
(ii) TWO uses of carboxylic acids. (5marks)
(b) State the medical use of Diphenhydramine amine salt. (2marks)

Question FIVE

(a) Giving one specific example, state FOUR uses of amines.

(8marks)

(b) (i) Explain why ionic compounds of amine salt have got excellent use in preparation of drugs .

(5marks)

(ii) State the medical use of Diphenhydramine amine salt.

(2marks)