



## TECHNICAL UNIVERSITY OF MOMBASA

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**Faculty of Engineering and Technology**  
**Department of Mechanical & Automotive Engineering**  
**UNIVERSITY EXAMINATION FOR:**  
**BSc. Mechanical Engineering**  
EMG 2302 : Engineering Thermodynamics II  
**END OF SEMESTER EXAMINATION**  
**SERIES: DECEMBER 2016**  
**TIME: 2 HOURS**  
**DATE: Pick Date Dec 2016**

**Instruction to Candidates:**

You should have the following for this examination

- *Answer booklet*
- *Non-Programmable scientific calculator*

This paper consists of **FIVE** questions. Attempt question **ONE** and any other **TWO** questions.

Maximum marks for each part of a question are as shown.

**Do not write on the question paper.**

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**Question One**

- a) Explain the following terms
- Available Energy
  - Unavailable Energy (4marks)
- b) Derive the equation for the decrease in available energy when heat is transferred through a finite temperature different. (9 marks)
- c) A system at 500K receives 7200kJ/min from a source at 1000K. The temperature of the atmosphere is 300K. Assuming that the temperatures of the system and source remain constant during heat transfer. Calculate
- The entropy produced during heat transfer

- ii. Decrease in available energy after heat transfer. (7 marks)

### Question Two

- a) Define the following terms
- Mole fraction
  - Partial pressure. (3 marks)
- b) 1 kg of air at 1 bar and 50°C is compressed adiabatically to a pressure of 5 bar. The air is then expanded at constant pressure to the initial volume. Determine work transfer and heat transfer for the whole path. (5 marks)
- c) A mixture of ideal gases consists of 2.5 kg of N<sub>2</sub> and 4.5 Kg of carbon dioxide at a pressure of 4 bar and a temperature of 25°C. Find
- The mole fraction of each constituent
  - Equivalent of molecular weight of the mixture
  - Equivalent gas constant of the mixture
  - Partial pressure and partial volumes
  - The volume and density of the mixture
  - C<sub>p</sub> and C<sub>v</sub> of the mixtures [take  $\gamma=1.285$  for CO<sub>2</sub> and  $\gamma=1.4$  for N<sub>2</sub>] (12 marks)

### Question Three

- a) Define the following terms
- Dry air
  - Saturated air
  - Wet-bulb temperature
  - Dry- bulb temperature (8 marks)
- b) Explain the adiabatic saturation process. (6 marks)
- c) The air supplied to a room of a building in cold season is to be at 17°C and a relative humidity of 60%. If the barometric pressure is 1.01325 bar. Calculate
- Specific humidity.
  - The dew point under these conditions. (6 marks)

### Question Four

- a) Describe a regenerative cycle with a single feed water heater and show its efficiency (12 marks)

- b) In a single-heater regenerative cycle the steam enters the turbine at 30 bar, 400°C and the exhaust pressure is 0.10 bar. The feed water heater is a direct contact type which operates at 5 bar. Find
- The efficiency and the steam rate of the cycle.
  - The increase in mean temperature of heat addition, efficiency and steam rate as compared to the Rankine cycle (without regeneration). Pump work may be neglected.

**Question Five**

- a) Explain the various operations of Rankine cycle and derive its efficiency. (10 marks)
- b) In Rankine cycle, the steam at inlet to turbine is saturated at a pressure of 35 bar and the exhaust pressure is 0.2 bar. Calculate
- The pump
  - Turbine work
  - Rankine efficiency
  - Condenser heat flow
  - The dryness at the of expansion. ( 10 marks)