



TECHNICAL UNIVERSITY OF MOMBASA

FACULTY OF ENGINEERING AND TECHNOLOGY

DEPARTMENT OF MECHANICAL & AUTOMOTIVE ENGINEERING

UNIVERSITY EXAMINATION FOR:

**THE DEGREE IN BACHELOR OF TECHNOLOGY IN MECHANICAL
ENGINEERING**

TMC 4226 : Engineering Thermodynamics I

END OF SEMESTER EXAMINATION

SERIES: AUGUST 2017

TIME: 2 HOURS

DATE: Pick Date Aug 2017

Instructions to Candidates

You should have the following for this examination

-Answer Booklet, examination pass and student ID

- Tables of 'Thermodynamic and Transport Properties of Fluids' by G.F.C Rodgers and Y.R Mayhew

This paper consists of **FIVE** questions. Attempt any **THREE** questions.

Do not write on the question paper.

Question One

- a) What is an Ideal gas (2 marks)
- b) State and explain the following
- i. Charle's Law
 - ii. Boyle's Law
 - iii. Avogadro's Law (9marks)
- c) A gas occupies a volume of 0.1m^3 at a temperature of 20°C and a pressure of 1.5 bar. Find the final temperature of the gas, if is compressed to a pressure of 7.5 bar and occupies a volume of 0.04 m^3 . (3 marks)

- d) A quantity of gas has a pressure of 350 KN/m^2 when its volume is 0.03 m^3 and its temperature is 35°C . If the value of $R=0.29 \text{ KJ/kg K}$.
- Calculate the mass of the gas .
 - If the gas pressure is increased to 1.05 MN/m^2 while the volume remains constant ,Calculate new temperature. (6 marks)

Question Two

- State the First law of thermodynamics. (2 marks)
- Derive from first principles the Steady flow energy Equation (10 marks)
- Air enters a gas turbine system with a velocity of 1005 m/s and has a specific volume of $0.9 \text{ m}^3/\text{kg}$. The inlet area of the gas turbine system is 0.06 m^2 . At exit the air has velocity of 140 m/s and has a specific volume of $1.4 \text{ m}^3/\text{kg}$, the specific enthalpy of the air is reduced by 140 KJ/kg and the air also has a heat transfer loss of 30 KJ/kg .

Determine:-

- The mass flow rate of the air through the turbine system in kg/s
- The power developed by the turbine system in kW (8 marks)

Question Three

- State Kelvin-Planck and Clausius statements of second Law of thermodynamics (4 marks)
- Differentiate between heat engine and reversed heat engine. (5 marks)
- Derive the efficiency equation of a heat engine. (4 marks)
- A MPUC students claims to have developed a heat engine with specifications as follows:

Power Developed = 76 kW

Fuel Burnt 4.8 kg/hr

Heating value of fuel = 7300 kJ/kg

Temperature limits = 980 k and 393 k

Evaluate if the student claim is valid?

(7marks)

Question Four

- a) State the properties of steam which make it a preferred working fluid. (3 marks)
- b) Explain the following terms
 - i. Wet steam
 - ii. Dry saturated steam
 - iii. Superheated steam (6 marks)
- c) Define dryness fraction of steam (2 marks)
- d) State advantages of superheated steam (4 marks)
- e) Calculate the quantity of heat required to produce 1 kg of steam at a pressure of 6 bar at a temperature of 250°C, under the following conditions. (steam tables at 6 bar $h_f=6704.4\text{kJ/kg}$, $h_{fg}=2085\text{kJ/kg}$, $t=158.8^\circ\text{C}$)
 - i. When the steam is wet having a dryness fraction of 0.9
 - ii. When the steam is dry saturated
 - iii. When it is superheated at a constant pressure of 250°C assuming the mean specific heat of superheated steam to be 2.3 kJ/kg K (8 marks)

Question Five

- (a) Define the term fuel. (2 mark)
- (b) Explain the advantages and disadvantages of liquid fuels over solid fuels. (5 marks)
- (c) A sample of coal has the following composition by mass.
Carbon 75%; Hydrogen 6%; Oxygen 8%, Nitrogen 2.5%; Sulphur 1.5% and ash %
Calculate the higher and lower calorific values per kg of coal (4 marks)
- (d) A blast furnace gas has the following volumetric composition:
 $\text{CO}_2=11\%$, $\text{CO} = 27\%$, $\text{H}_2 = 2\%$ and $\text{N}_2 = 60\%$

Calculate

- (i) The theoretical volume of air required for the complete combustion of 1m^3 of the gas
- (ii) The percentage composition of dry flue gasses by volume

(Assume that air contains 21% of O_2 and 79% of N_2 by volume) (9 marks)