

# TECHNICAL UNIVERSITY OF MOMBASA

# FACULTY OF APPLIED AND HEALTH SCIENCES DEPARTMENT OF PURE & APPLIED SCIENCES

# **UNIVERSITY EXAMINATION FOR:**

# **BTAC**

ACH 4307: MATERIAL CHEMISTRY END OF SEMESTER EXAMINATION

**SERIES: Special/Supplementary Examinations** 

TIME: 2 HOURS

**DATE:** Pick Date Select Month 2017

# **Instructions to Candidates**

You should have the following for this examination

-Answer Booklet, examination pass and student ID

This paper consists of **FIVE** questions. Answer question ONE (Compulsory) and any other TWO questions. **Do not write on the question paper.** 

#### **Ouestion ONE**

- (a) Tabulate the seven crystal systems giving the restrictions of axes and angles (9 marks)
- (b) What is:
  - (i) A biomaterial (1 mark)
  - (ii) Biocompatibility (1 mark)
- (c) What is a laser? (1 Mark)

What is the structure of semiconducting laser? (1 mark)

- (d) How does the Band Theory help us understand what makes conductor, insulator, or semiconductor? (4 marks)
- (e) What is:
  - (i) A Superconductor? (2 marks)
  - (ii) The Critical Temperature for Superconductors (3 marks)
- (f) What kind of information do the following analytical techniques give:
  - (i) X-Ray Diffraction
  - (ii) Neutron Diffraction

- (iii) Solid State NMR
- (iv) Scanning Electron Microscopy
- (v) Transmission Electron Microscopy
- (vi) Elemental dispersion analyzer using X-rays
- (vii) IR Spectroscopy
- (viii) X-ray fluorescence spectroscopy (1 mark each)

# **Question TWO**

With the aid of diagrams, describe the 14 Bravais lattices (20 marks)

# **Question THREE**

- (a) Give a comparison of the properties of liquid crystal and other phases of matter (4 marks)
- (b) Describe with the aid of diagrams the characteristics and type of ordering in the following phases of liquid-crystals:
  - (i) Nematic phase (4 marks)
  - (ii) Smectic phase (4 marks)
  - (iii) Cholesteric phase (Chiral nematic phase) (4 marks)
  - (iv) Discotic phase (4 marks)

# **Question FOUR**

Write notes on Zeolites with reference to:

- (a) Mineralogy (5 marks)
- (b) Common properties (7 marks)
- (c) Describe:
  - (i) The manufacture of ZeoPro from a natural source of clinoptilolite (4 marks)
  - (ii) How ZeoPro works (4 marks)

# **Question FIVE**

Tabulate the physical and structural properties associated with: Ionic; Covalent; Metallic; and van der Waals bonds (20 marks)