

Technical University of Mombasa

Faculty of Applied and Health Sciences

DEPARTMENT OF MEDICAL SCIENCES

UNIVERSITY EXAMINATION FOR THE DEGREE OF BACHELOR OF MEDICAL LABORATORY SCIENCES

BMLS 13M REGULAR

AML 4201: CLINICAL CHEMISTRY I

SEMESTER EXAMINATION

APRIL 2014 SERIES

2

HOURS

Instructions to candidates:

This paper consists of **TWO** sections **A** and **B Section A** -Contains MCQS, Answer **ALL** questions in **Section B**.

SECTION A - MCQs - (30marks)

- 1. Water constitutes the following percentage of body weight
 - a) 80%
 - b) 95%
 - c) 75%
 - d) 60%
- 2. The most ideal container for random urine specimen is
 - a) 10 litre
 - b) Bijou bottle
 - c) McCartney bottle
 - d) Winchester bottle
 - e) Universal bottle
- 3. Urine preservatives include the following except
 - a) Boric acid

- b) HNO₃
- c) Sodium Carbonate
- d) Concentrated HCL
- e) Sodium Nitroprosside
- 4. What is the term given to an increase in urine volume
 - a) Anuria
 - b) Renal dysfunction
 - c) Oliguria
 - d) Polyuria
 - e) Retinopathy
- 5. What is the term given to absence in urinary volume
 - a) Anuria
 - b) Renal dysfunction
 - c) Oliguria
 - d) Polyuria
 - e) Retinopathy
- 6. Which one of the following can be used to determine specific growing of urine
 - a) Refractometer
 - b) Osmometer
 - c) Ion selective electrode analyzer
 - d) Centrifuge
 - e) Micropipette
- 7. A blood pH of 7.35 would indicate:
 - a) Normal
 - b) Acidosis
 - c) Alkalosis
 - d) Neutral
 - e) Extremely alkaline
- 8. Which of the following would occur from an increased capillary hydrostatic pressure?
 - a) Dehydration
 - b) Hypotonic hydration
 - c) Edema
 - d) Diabetes mellitus
 - e) Diabetes insipidus
- 9. What is the major cation in the ECF?
 - a) Chloride
 - b) Sodium
 - c) Bicarbanate
 - d) Potassium
 - e) Zinc chloride
- 10. Which of the following compartments has the highest proportion water?

- a) Intracellular
- b) Trancellar fluid compartment
- c) Intravascular fluid compartment
- d) Interstitial fluid compartment
- e) None of the above
- 11. What is the hormone which controls glycogenolysis?
 - a) Aldosterone
 - b) ADH
 - c) Insulin
 - d) Glucagon
 - e) Growth hormone
- 12. Match the following with the correct chemical formula
 - a) H_2CO_3 = weak acid
 - b) Lactic acid = strong acid
 - c) Sulphuric acid = Weak acid
 - d) Potassium hydroxide = weak base
 - e) HCL = Weak acid
- 13. Metabolic alkalosis is seen in
 - a) An increase in rate and depth of breath
 - b) A decrease in rate and depth of breath
 - c) Excessive administration of sodium bicarbonate
 - d) Administration of substances whose metabolism produces H+
 - e) Anaemia
- 14. The normal range for sodium in blood is
 - a) 135 145 mmol/L
 - b) 3.8 5.0 mmol/L
 - c) 6.8 10.0 mmol/L
 - d) 5.0 6.8 mmol/L
 - e) 10-15mmol/L
- 15. Which of the following does not provide buffering power of blood
 - a) HCO₃/H₂CO₃
 - b) Na+
 - c) Phosphate ions
 - d) Plasma proteins
 - e) None of the above
- 16. A blood pH of 7.45 would indicate
 - a) Normal
 - b) Acidosis
 - c) Alkalosis
 - d) Neutral
 - e) Extreme acidosis

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a) b)	ow does alcohol function as a diuretic? Alcohol dilutes the blood and thus endoles increased urine output Alcohol receptors in the liver sense its presence and trigger a biochemical pathw urine output to rid the body of alcohol	vay that increase
a) b) c) d)	Are anions of the intracellular fluids Are anions of extracellular fluid Have a normal plasma range of 150 – 160mmol/L Are cations of the extra cellular fluid None of the above	
a)b)c)d)	at is the name of the condition of High Na ⁺ levels in the ECF? Hyper kalemia Hypokalemia Hypernatrema Hyponatremia Hyperchloremia	
a)b)c)d)	troprusside in the Berthelot reaction functions as a Buffer Catalyst Substrate Chromogen Electron donor	
a)b)c)d)	ich of the following is the end product of protein metabolism Bilirubin Urea Uric acid Creative None of the above	
c)	During increased respiration over the long term, more hemoglobin is produced the buffering of blood During increased respiration over the long term more lactic is produced It is not possible	thus increasing

18. How is it possible for the rate and depth of breathing to affect hydrogen ion concentration in body

a) During increased air exchange, more oxygen is exchanged with body cells binding hydrogen ions

17. Fluid bathing tissue is classified specifically as _____

a) Intracellularb) Extracellularc) Interstitiald) Intracascolare) Intradermal

- c) Alcohol inhibits the release of ADH and flues urine out pot is increase
- d) Alcohol increases the heart pulse rate which then increases urine output
- e) None of the above
- 24. Which body compartment contains the lowest relative amount of water?
 - a) Extracellulant
 - b) Intracellular
 - c) Plasma
 - d) Trancellular
 - e) None of the above
- 25. Which of these conditions leads to a severe threatening water loss
 - a) Thalasemia
 - b) Diabetes insipidus
 - c) Hyponatraemia
 - d) Coronary heart disease
- 26. Which of the following is true regarding diabetes mellitus
 - a) IDDM is community reach in persons above 40 years
 - b) NIDDM tends to occur in obese people
 - c) Serum insulin level is absolutely absent in NIDDM
 - d) Serum protein lends are elevated
 - e) None of the above
- 27. Which of the following is true of insulin
 - a) Is a protein synthesized by β cells of islets of langerhan
 - b) Secretion is stimulated by dietery protein
 - c) Increases gluconeogenesis in liver
 - d) Increase glycogenolysis in liver
 - e) Is an enzyme
- 28. Which of the following conditions is associated with pre-renal elevated urea levels
 - a) Severe diarrhea
 - b) Jaundice
 - c) Alomersulonephritis
 - d) Nephrotic syndrome
 - e) Renal tumours
- 29. Which of the following disease conditions is associated with elevated blood uric acid levels
 - a) Diabetes
 - b) Gout
 - c) Malaria
 - d) Syphilis
 - e) Anaemia
- 30. Casts are formed
 - a) If urine is left to stand for long
 - b) In the urinary bladder

- c) In renal tubules
- d) In the liver
- e) In liver dysfunction

SECTION B ESSAY (ANSWER ALL QUESTION IN THIS SECTION)

1. (a) Describe the disease conditions associated with pre-renal increases in blood urea

(10

marks)

(b) Describe respiratory alkalosis and metabolic alkalosis (10 marks)

2. (a) Describe the imbalances associated with Na+ in blood (10 marks)

(b) Describe the oral glucose tolerance test (10 marks)