

# Technical University of Mombasa

# Faculty of Applied and Health Sciences

## **DEPARTMENT OF PURE AND APPLIED SCIENCES**

UNIVERSITY EXAMINATION FOR THE DEGREE OF BACHELOR OF TECHNOLOGY IN APPLIED CHEMISTRY

# BTAC 11M

# ACH 4305: INDUSTRIAL PROCESSES

## SPECIAL/SUPPLEMENTARY EXAMINATION

OCTOBER 2013 SERIES 2 HOURS Instructions to candidates:

This paper consist of **FIVE** questions Answer question **ONE** (compulsory) and any other **TWO** questions

#### **Question ONE**

- a) State be chatdiers principle as applied to a chemical system when there is a change in the following:-
  - (i) Temperature
  - (ii) Pressure

b) Outline mechanical pulping (5marks) (6marks)

c) State FOUR raw materials for manufacture of ordinary glass. (4marks)

- d) State :-
  - (i) Three conditions imposed in convening zone of kiln which govern the quality of the cement clinker. (3marks)
  - (ii) Advantage of silicate cements over Portland cements (2marks)
  - (iii) Differentiate between green liquor and white liquor (4marks)
  - (iv) List TWO types of furnaces used in glass manufacture and state when each is used (4marks)
  - (v) Name an important by product from black liquor chemical recovery and state its use in paper manufacture. (2marks)

#### **Question TWO**

a)	State TWO types of materials necessary for cement manufacture.	(2marks)
b)	Describe cement manufacture by the wet process	(5marks)

- c) Name the FOUR clinker bogue compounds responsible for cement properties and state the property each is responsible for. (6marks)
- d) State the formulae for the three control factor used in the cement manufacture (7marks)

#### **Question THREE**

- a) List any THREE sources of raw materials for pulp and paper manufacture (3marks)
- b) State the functions of the following paper additives
  - (I) Rosin emulsion
  - (II) Dyes (2marks)
- c) State FOUR properties of the fillers that are used in paper manufacture (4marks)
- d) Figure I is a simplified flow diagram for pulp production using magnesium hydrogen sulphite.
  - (i) Identify stream labelled P and Q, and the units labelled 1,2,3 and 4
  - (ii) State the advantage of using magnesium hydrogen sulphite over calcium hydrogen sulphite in sulphate pulping.
  - (iii) Outline the next treatments the pulp undergoes to produce a relatively pure pulp

#### **Question FOUR**

Assume you are a newly employed chemical process engineer at a proposed sulphuric acid manufacturing plant on examining the design data, you come across the following reaction that has been proposed hydration of sulphur-trioxide to produce sulphuric acid.

 $SO_3 + H_2O \rightarrow H_2SO_4 \Delta H - 200 K Jmol^{-1}$ 

a)	(i)	Would you agree or disagree with this proposal	(1mark)
	(ii)	Give reasons for your opinion of (i) above	(3marks)
	(iii)	From your knowledge of the manufacture of sulphuric acid, w would you give?	what alternative (4marks)
b)	State any FOUR reason why deodorized ethanol is considered a suitable vehicle in perfumes (4marks)		
c)		e fertilizer and from your knowledge of salts, differentiate betweenic fertilizers.	een organic and (4marks)
d)	Define	e the term "Broke" as used in the paper manufacture	(2marks)
e)	Explai	in TWO types of broke	(2marks)

#### **Question FIVE**

- a) List THREE sulphide ores used in sulphuric acid manufacture. (3marks)
- b) For the reversible reaction which is in equilibrium

 $PCl_{5}(g) \longrightarrow PCl_{3}(g) + Cl_{2}(g) \Delta H = + 124KJ/Mol$ 

- (i) State THREE ways by which the rate of production of  $Cl_2$  is enhanced. (3marks)
- (ii) Explain the effect of the following on the reaction.
  - Doubling partial pressure of gas PCl<sub>5</sub>
  - Doubling the size of the reaction vessel
  - Adding an inert gas
  - Increasing the reaction temperature
  - Increasing the reaction temperature
  - Withdrawing product gases PCl<sub>3</sub> and Cl<sub>2</sub>.
  - Doubling the concentration of the reactant gas. (8marks)
- (iii) Wet ammonium nitrate granules were found to contain 60% by wt water. After

drying it was found that 80 % of the original water was removed. Calculate the water solid composition by weight of the dried salts. (6marks)