



TECHNICAL UNIVERSITY OF MOMBASA
Faculty of Applied & Health
Sciences

DEPARTMENT OF MATHEMATICS & PHYSICS

CERTIFICATE IN BUILDING & CIVIL ENGINEERING (CBCE 13S)

APS 1101: PHYSICAL SCIENCE FOR ENGINEERS

END OF SEMESTER EXAMINATION

SERIES: DECEMBER 2013

TIME ALLOWED: 2 HOURS

Instructions to Candidates:

You should have the following for this examination

- *Answer Booklet*

This paper consist of **FIVE** questions

Answer question **ONE (COMPULSORY)** and any other **TWO** questions

Maximum marks for each part of a question are as shown

This paper consists of **FOUR** printed pages

Question One (Compulsory)

- a) State the Kirchhoff's Law on:
(i) Current
(ii) Voltage (4 marks)
- b) With the aid of symbolic diagrams, explain the following with regard to diodes:
(i) Forward biasing
(ii) Reverse biasing (4 marks)
- c) Describe **FOUR** factors that affect resistance of a conductor (8 marks)
- d) Explain the following:
(i) Rectification
(ii) Voltage regulation (4 marks)

Question Two

- a) Given the colour codes of the following resistors, determine the value of the resistors and calculate their maximum and minimum values in Kilo-ohm ($K \Omega$) (8 marks)
- (i) Red, Red, Green, Gold
(ii) Blue, Black, Red, Silver
(iii) Red, Blue, Yellow, Red
(iv) Green, Yellow, Green
- b) Explain how the following chemical bonds are formed.
(i) Covalent bond
(ii) Ionic bond
(iii) Molecular bond (6 marks)
- c) Find amount of C_2H_6 produced using 0.3moles C_2H_2 and 0.4 moles H_2 using the following chemical reaction:
$$C_2H_{2(g)} + 2H_{2(g)} \rightarrow C_2H_{6(g)}$$
 (6 marks)

Question Three

- a) A circuit Network consisting of capacitors is shown:
Calculate:
(i) Total effective capacitance
(ii) Total charge stored in the circuit
(iii) Electrical energy stored in the circuit (6 marks)
- b) State **FOUR** factors that affect the resistance of a wire (4 marks)

- c) State the following:
- (i) Kirchoff's Law on voltage
 - (ii) Kirchoff's Law on current (4 marks)
- d) Calculate the length of a steel wire of 0.8mm diameter having resistance of $192 \frac{\Omega}{\text{m}}$ (for steel = $0.013 \times 10^{-6} \text{ m}$) (3 marks)
- e) Explain the **THREE** transformer Losses and state how they can be minimized. (3 marks)

Question Four

- a) Three resistors of $120 \frac{\Omega}{\text{m}}$, $50 \frac{\Omega}{\text{m}}$ and $70 \frac{\Omega}{\text{m}}$ are connected in parallel and then connected in series to a $100 \frac{\Omega}{\text{m}}$ resistor. The circuit is supplied with 15V D.C. Determine:
- (i) Total current in the circuit
 - (ii) Current through the $50 \frac{\Omega}{\text{m}}$ resistor
 - (iii) Voltage drop across the $100 \frac{\Omega}{\text{m}}$ resistor
 - (iv) Total power dissipated in the circuit (8 marks)
- b) Distinguish between A.C and D.C energy citing at least two application of each. (6 marks)
- c) With the aid of a circuit diagrams and wave form diagrams. Explain the operation of a full-wave bridge rectifier (6 marks)

Question Five

- a) The grid below represents part of the periodic table. Study it and answer the questions that follow. The letters do not represent the actual symbols of the elements. (1 mark)

A							K
B	D				G	I	
			F		H		L
C	E					J	

- (i) Which letter represents an element which is least reactive? (1 mark)
 - (ii) Why are elements D and E referred to as Alkaline Earth Metals (1 mark)
- b) (i) How are the atomic radius of F and H compare (2 marks)
- (ii) Select two letters representing a pair of element that would react most expossively (2 marks)
- (iii) Write an equation showing how D forms its ions (2 marks)

- c) (i) Write the formulae of Bromide of D (1 mark)
(ii) Write the formulae of sulphate of C (1 mark)
- d) What type of bonding exists between:
(i) E and I
(ii) G and J (2 marks)
- e) (i) Explain why the melting point of J is higher than that of I (2 marks)
(ii) Apart from the decrease in energy levels, explain the difference between 1st and 2nd ionization energies (2 marks)
- f) Using symbolic diagram, differentiate between forward biasing and reverse biasing a crystal diode. (4 marks)